

Enantiomerically pure 1,2,5-triphenylphospholane through the synthesis and resolution of the chiral *trans*-(2,5)-diphenylphospholanic acid

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Abstract

The synthesis and resolution of *trans*-(2,5)-diphenylphospholanic acid 7 is described. The phosphinic acid 7 was converted into optically active (1,2,5)-triphenylphospholane 5 which was used as a chiral ligand in Rh-catalyzed hydrogenation of *N*-acetyl dehydrophenylalanine methyl ester to give quantitative yield of methyl N-acetylphenylalaninate with 82 % e.e. © 1999 Elsevier Science Ltd. All rights reserved.

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Asymmetric hydrogenations, hydrosilylations and C-C bond forming reactions are well developed using homogeneous catalysis with transition metal complexes bearing diphosphines as chiral ligands [1,2]. Among the recently devised chiral diphosphine ligands, a number incorporate the phospholanyl skeleton as a common structural motif [3-11]. Noteably, Burk synthesized [4,5] and developed the use [6] of 1,2-bis(2,5-dialkylphospholanyl)benzene, (DuPHOS) and 1,2-bis(2,5-dialkylphospholanyl)ethane (BPE) which incorporate the *trans*-2,5-disubstituted phospholanyl framework 1 (scheme 1). Since then a number of diphosphines have been described wich have two linked (2,5-dimethylphospholanyl) fragments of the same configuration [10,11].

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Aiming at the building of the trans-2,5-diphenylphospholanyl framework (1, R = Ph), we found that Burk's procedure [4,5] (scheme 1) was not convenient as both the cyclic sulfate 2 (R=Ph) and dimesylate 3 (R=Ph) suffered elimination on treatment with PhPH₂ / BuLi to afford the 1,4-diphenylbuta-1,3-diene.

Scheme 1. Synthesis of DuPHOS and BPE according to Burk (R =Me, Et, Pr, i-Pr)

We previously reported the synthesis of $trans-(2R^*,5R^*)-1-oxo-1,2,5$ -triphenylphospholane **4**, its resolution and subsequent reduction to optically active phospholane **5** by a different synthetic scheme [12]. However, as the resolution procedure (preparative chiral liquid chromatography) of **4** was not practical on a gram scale, we looked for an easier access to enantiomerically enriched **4**. As the asymmetric sparteine-lithium promoted deprotonation / acetic acid protonation sequence of the $meso\ r-1-oxo-1,t-2,t-5$ -triphenylphospholane **6** could only produce optically active material **5** with a maximum of 45 % e.e. [13], we turned to the synthesis of the corresponding 1-hydroxy-r-1-oxo-c-2,t-5-diphenylphospholane **7** and its resolution through crystallization of diastereomeric salts.

We now report the synthesis and resolution of trans-(2,5)-diphenylphospholanic acid 7, its conversion to optically active $(2R^*,5R^*)$ -(1,2,5)-triphenylphospholane-1-oxide 4 and the reduction of 4 to give the corresponding phosphine 5.

Reaction of commercially available (*N*,*N*-diisopropylamino)dichlorophosphine with 1,4-diphenylbuta-1,3-diene afforded 1-(*N*,*N*-dimethylamino)-*r*-1-oxo-*t*-2,*t*-5-triphenyl-phosphol-3-ene which was hydrogenated to give the corresponding phospholane 8 (scheme 2). Isomerization of 8 to the more stable *trans* isomer 9 was carried out in methanol, with an excess of sodium methoxide. Phospholanic acid 7 was obtained by the acid-promoted hydrolysis of amide 9 and was readily resolved by crystallization of the diastereomeric quinine salts¹.

¹ The enantiomeric purity of 7 could be checked as its methyl ester by chiral chromatography (Regis[®] (S,S)-Whelk 01 column with dichloromethane/n-hexane/isopropanol (5/4/1) as eluent). Selected data for (-)-7: mp 269-270 °C. [α]_D²⁵ = -102.7 (c = 0.6, CH₂Cl₂). ¹H NMR (CD₃OD): 7.2-7.0 (10H, m) 3.2-3.0 (2H, m) 2.4-1.9 (2H, m). ¹³C NMR (CD₃OD): 138.0 (d, J = 5.7 Hz) 129.7 (d, J = 5.5 Hz) 129.5 (d, J = 1.9 Hz) 127.7 (d, J = 2.4 Hz) 47.0 (d, J = 87.0 Hz) 30.2 (d, J = 11.9 Hz). ³¹P NMR (CD₃OD): 66.0 Anal. Calcd for C₁₆H₁₇O₂P: C, 70.58; H, 6.29; P, 11.38. Found: C, 70.48; H, 6.33; P, 11.08.

i) (CH₃)₂NPCl₂, AlCl₃, CH₂Cl₂, -10°C, then NaHCO₃ / EDTA, 0°C ii) 5% Pd / C, H₂ (30-50 atm.), CH₂Cl₂, 16 h, 65%(two steps). iii) MeONa (5 equiv.), MeOH, amb. temp., 16 h, quant. iv) aq. HCl, MeOH; v) quinine, 1 equiv., separation of diastereomeric salts, 40%

Scheme 2. Synthesis of the trans-(2,5)-diphenylphospholanic acid.

1-Oxo-1,2,5-triphenylphospholane 4 was obtained by one of two different procedures from chloride 10 (scheme 3), either by coupling with the diphenyl lithiocuprate, or via reduction to the secondary phosphine oxide 11 followed by a palladium-catalyzed coupling with phenyl iodide. The stereochemical integrity (diastereomeric and enantiomeric purity) of 4 could be checked by chiral HPLC analysis². The oxide 4 was converted to the air-stable borane complex 12 of the optically active *trans*-(2,5)-diphenylphospholane 5 through a reduction / complexation sequence. The free phosphine 5 was regenerated from 12 by reaction with half an equivalent of DABCO in toluene.

i) (COCl)₂, THF, quant.; ii) Ph₂CuLi, - 78°C, THF, 65 %; iii) DiBAl-H, 63%; iv) PhI, iPr₂NEt, 5 mol% [Pd(dba)₂], 7.5 mol% dppp, 88 %; v) LiAlH₄, CeCl₃, then BH₃-THF 73 %; vi) DABCO, toluene, quant.

Scheme 3. Synthesis of the optically active (1,2,5)-triphenylphospholane.

The phospholane 5 was examined as a chiral ligand in rhodium-catalyzed hydrogenation of N-acetyl dehydrophenylalanine methyl ester 13. Hydrogenation at atmospheric pressure using 1 mol % of $[RhCl(cod)]_2$ and 2.1 mol % (+)-5 (in situ catalyst) gave (S)-14 in quantitative yield and 82 % e.e., as determined by chiral hplc.

² Daicel[®] Chiralcel OD-H column with *n*-hexane/isopropanol (3/1) as eluent

Ph NHAc H2 (1 Atm), [cat*] Ph NHAc
$$CO_2Me$$
 13 (S)-14, quant. (82% ee) $[cat^*] = [Rh(COD)CI]_2 + 2.1 \text{ eq. (+)-5 / Rh}$

This result compares favorably with those obtained using P-chiral monophosphines (85 % ee in the Rh-catalyzed hydrogenation of the corresponding acid with o-anisylcyclohexylmethylphosphine as the ligand) [14] and other phospholanes (60% ee in the Rh-catalyzed hydrogenation of 13 with trans-(2,5)-dimethyl-1-phenylphospholane 2 (R = Me)) [4].

Work is in progress to apply the synthetic procedures to other monophosphines and diphosphines containing the chiral trans-2,5-diphenylphospholanyl framework and evaluate them as chiral ligands in various enantioselective metal-catalyzed reactions.

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References

- [1] Kagan HB. Chiral Ligands for Asymmetric Catalysis. In: Morrison JD, Editor. Asymmetric Synthesis. Orlando: Academic Press, 1985; vol. 5, chapter 1.
- [2] Brunner H, Zettlmeier W. Handbook of Enantioselective Catalysis with Transition Metal Compounds, vol. I and II, Weinheim: VCH, 1993.
- [3] Brunner H, Sievi R. J. Organomet. Chem. 1987, 328, 71-80
- [4] Burk MJ, Feaster JE, Harlow RL. Tetrahedron: Asymmetry 1991, 2, 569-592.
- [5] Burk MJ, Feaster JE, Nugent WA, Harlow RL. J. Am. Chem. Soc. 1993, 115, 10125-10138.
- [6] Burk MJ, Wang YM, Lee JR. J. Am. Chem. Soc. 1996, 118, 5142-5143, and references cited therein.
- [7] Longeau A, Durand S, Spiegel A, Knochel P. Tetrahedron: Asymmetry 1997, 8, 987-990
- [8] Holz J, Quirmbach M, Schmidt U, Heller D, Stürmer R, Börner A. J. Org. Chem. 1998, 63, 8031-8034.
- [9] Jiang Q. Jiang Y, Xiao D, Cao P, Zhang X. Angew. Chem. Int. Ed. 1998, 37, 1100-1003.
- [10] Morimoto T, Ando N, Achiwa K. Synlett, 1996, 1211-1212.
- [11] Dierkes P, Ramdeehul S, Barloy L, De Cian A, Fischer J, Kamer PCJ, van Leeuwen PWNM, Osborn JA. Angew. Chem. Int. Ed. 1998, 37, 3116-3118.
- [12] Fiaud J-C, Legros J-Y. Tetrahedron Lett. 1991, 32, 5089-5092.
- [13] Guillen F, Moinet C, Fiaud J-C. Bull. Soc. Chim. Fr 1997, 134, 371-374.
- [14] Knowles WS, Sabacky MJ, Vineyard BD. J. Chem. Soc. Chem. Commun. 1972, 10-11.